

Lampiran 1.1 Perhitungan uji aktivitas antioksidan

a. Pembuatan Larutan induk DPPH

Konsentrasi Larutan DPPH 5 mg dalam 50 ml

$$\text{Konsentrasi} = 5\text{mg}$$

$$\text{Volume} = 50 \text{ ml} = 0,05 \text{ L}$$

$$\text{Ppm} = \frac{\text{mg}}{\text{L}}$$

$$x = \frac{5\text{mg}}{0,005 \text{ L}}$$

$$x = 100 \text{ mg/L}$$

b. Penyiapan larutan stok sampel uji (1000 ppm)

Larutan stok sampel uji = 50 mg

$$\text{Volume} = 50 \text{ ml} = 0,05 \text{ liter}$$

$$\text{Ppm} = \frac{\text{mg}}{\text{L}}$$

$$x = \frac{5\text{mg}}{0,005 \text{ L}}$$

$$x = 1000 \text{ mg/L}$$

c. Perhitungan pengenceran larutan stok sampel uji

Perhitungan pengenceran larutan stok ekstrak dan minyak atsiri daun cengkeh 1000 ppm dengan konsentrasi (10 ppm, 20 ppm, 30 ppm , 40 ppm , 50 ppm) dalam labu ukur 5 ml.

$$10 \text{ ppm} = \frac{10 \text{ ppm} \times 5 \text{ ml}}{1000 \text{ ppm}} = 0,05 \text{ ml}$$

$$20 \text{ ppm} = \frac{20 \text{ ppm} \times 5 \text{ ml}}{1000 \text{ ppm}} = 0,1 \text{ ml}$$

$$30 \text{ ppm} = \frac{30 \text{ ppm} \times 5 \text{ ml}}{1000 \text{ ppm}} = 0,15 \text{ ml}$$

$$40 \text{ ppm} = \frac{40 \text{ ppm} \times 5 \text{ ml}}{1000 \text{ ppm}} = 0,2 \text{ ml}$$

$$50 \text{ ppm} = \frac{50 \text{ ppm} \times 5 \text{ ml}}{1000 \text{ ppm}} = 0,25 \text{ ml}$$

d. Perhitungan nilai IC50 minyak atsiri daun cengkeh

Perhitungan nilai IC50 dengan konsentrasi (10 ppm, 20 ppm, 30 ppm , 40 ppm , 50 ppm)

$$10 \text{ ppm} = \frac{0,304 \times 0,074}{0,304} \times 100\%$$

$$= 75,657$$

$$20 \text{ ppm} = \frac{0,304 - 0,178}{0,304} \times 100\%$$

$$= 41,447$$

$$30 \text{ ppm} = \frac{0,304 - 0,18}{0,304} \times 100\%$$

$$= 40,789$$

$$40 \text{ ppm} = \frac{0,304 - 0}{0,304} \times 100\%$$

$$= 100$$

$$50 \text{ ppm} = \frac{0,304 \times 0,037}{0,304} \times 100\%$$

$$= 87,8228$$

$$y = 0,829 x + 44,276$$

$$50 = 0,829 x + 44,276$$

$$x = \frac{50 - 44,276}{0,829}$$

$$= \frac{5,724}{0,829}$$

$$= 6,904 < 50 \text{ (sangat kuat)}$$

e. Perhitungan nilai IC50 ekstrak daun cengkeh

Perhitungan nilai IC50 dengan konsentrasi (10 ppm, 20 ppm, 30 ppm , 40 ppm , 50 ppm)

$$10 \text{ ppm} = \frac{0,304 - (-0,056)}{0,304} \times 100\%$$

$$= 118,421$$

$$20 \text{ ppm} = \frac{0,304 - 0,142}{0,304} \times 100\%$$

$$= 53,289$$

$$30 \text{ ppm} = \frac{0,304 - 0,108}{0,304} \times 100\%$$

$$= 64,473$$

$$40\text{ppm} = \frac{0,304 - (-0,026)}{0,304} \times 100\%$$

$$= 108,552$$

$$50\text{ppm} = \frac{0,304 \times 0,037}{0,304} \times 100\%$$

$$= 87,8228$$

$$y = 0,2303 x + 82,488$$

$$50 = 0,2303 x + 823,488$$

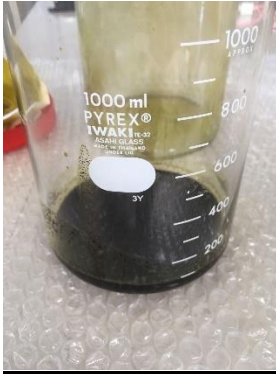
$$x = \frac{50 - 823,488}{0,2303}$$

$$= -141,068 < 50 \text{ (sangat kuat)}$$

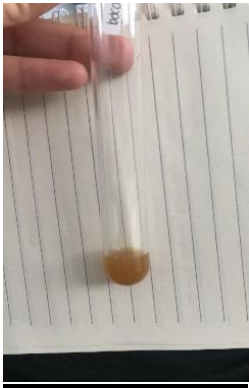
$$= 6,904 < 50 \text{ (sangat kuat)}$$

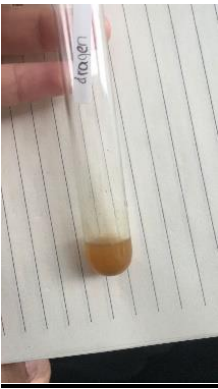
Lampiran

NO.	Gambar Sampel	Ket
1.	Daun cengkeh 	Pengeringan

2.	simplisia 	Daun cengkeh yang sudah di blender
3.	Ekstrak 	Hasil rotari
4.	<u>Meserasi</u> 	Meserasi

Lampiran

NO.	Gambar Sampel	Ket
1.	Alkaloid (Mayer) 	+
2.	Alkaloid (bochardat) 	+

3.	Alkaloid (dragendroff) 	+
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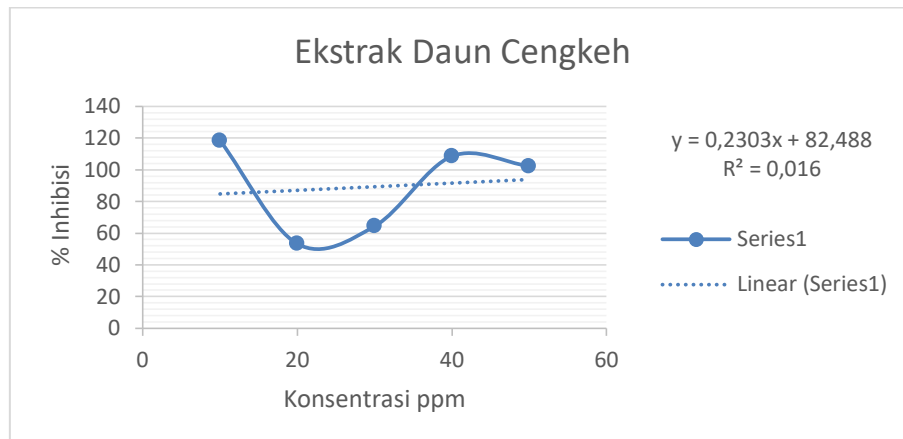
4.	<u>Flavonoid</u> 	+
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5.	Saponin 	+
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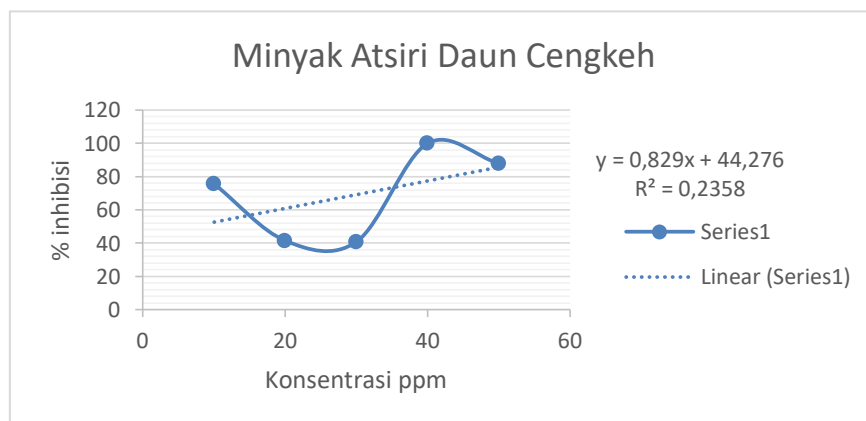
6.	Tanin 	+
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7.	Steroid 	+
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Lampiran 4.4 Kurva



Gambar 4.1 Grafik kurva regresi linear ekstrak daun cengkeh



Gambar 4.2 Grafik kurva regresi linear minyak atsiri

